# **Influence of fluorine doping on the transport**  properties of GdBa<sub>2</sub>Cu<sub>3</sub>O<sub>7-x</sub>

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Undoped and fluorine-doped materials of GdBa<sub>2</sub>Cu<sub>3</sub>O<sub>7-x</sub>F<sub>x</sub> (with  $x = 0$ , 0.05, 0.1 and 0.2) were prepared at high temperature  $(950^{\circ} C)$  by solid state reaction followed by oxygen treatment. X-ray diffractograms of the undoped sample showed single orthorhombic phase. **Resis**tivity measurements showed that  $T_c(0)$  decreases with fluorine concentration. Thermopower measurements indicated a peak just before the onset of the superconducting transition. The reasons for the decrease of  $T_c$  and the anomalous behaviour of thermopower have been explained.

## **1. Introduction**

Ever since the discovery of high-temperature superconductivity  $(T_c(0) = 30K)$  in the La-Ba-Cu-O system by Bednorz and Muller [1] and later 90K superconductivity in the Y-Ba-Cu-O system by Wu *et al.* [2], there has been intense research activity on these high  $T_c$  materials.

Several research groups replaced yttrium with rare earth atoms and they found that  $T_c$  was not altered very little. Similarly, several others added small amounts of 3-d ions (chromium, manganese, iron, nickel, cobalt or zirconium) with the intention of replacing the copper ion. These doped compounds always showed a decrease in  $T_c$ . Several other workers tried to replace oxygen with fluorine, selenium, etc. To obtain higher  $T_c$ , Ovshinsky *et al.* [3] reported  $T_c$  at 155 K by partial substitution of fluorine for oxygen. Radhakrishnan *et al.* [4] added BaF, as a dopant in  $YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7</sub>$  compound and their results showed that fluorine did not enter the lattice. They also reported that there was no significant change in  $T_c$ . In the fluorine-implanted YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7</sub> system, Narliker [5] observed a sudden drop in resistance to a very small value around 280 to 290 K; it then remained constant to 89 K and then dropped to zero. Meng Xian-Ren *et al.* [6] also observed zero resistance at 148.5K in fluorine-implanted Y-Ba-Cu-O compound and this they attributed to a new superconducting phase. They also found that the new phase was quite unstable and it disappeared after three or more thermal cycles from room temperature (RT) to liquid nitrogen temperature (LNT). All these results suggest that fluorine plays an active role in these materials but the problem appears to be the stability of the new superconducting phase. All these studies have been confined to the Y-Ba-Cu-O system. We have studied the influence of fluorine on resistivity and thermopower of  $GdBa_2Cu_3O_{7-x}$  material and the results are presented. It was observed that  $T_c(0)$  gradually decreases with fluorine concentration and thermopower measurements indicate anomalous behaviour just before the onset of superconductivity. A possible explanation for these observations is given.

## **2. Experimental procedure**

Undoped and fluorine-doped  $GdBa_2Cu_3O_{7-x}F_x$  (with  $x = 0, 0.05, 0.1$  and 0.2) compounds were synthesized by taking appropriate amounts of  $Gd_2O_3$ ,  $GdF_3$ ,  $BaCO<sub>3</sub>$  and CuO, such that gadolinium, barium and copper were in the molar ratio of  $1:2:3$ . The ground compounds were placed in a platinum crucible and reacted in air at high temperature (nearly  $950^{\circ}$  C) for 24 h. The compound so formed was cooled slowly and the process repeated. The material was powdered and pelletized by applying 4 to 5 tonne pressure. The pellets were then heated at  $950^{\circ}$ C in flowing oxygen and slowly cooled to room temperature. X-ray diffractograms of all the samples so prepared were recorded with a Phillips X-ray diffractometer employing  $CuK\alpha$ radiation. Resistivity measurements were made by a four-probe technique in a glass cryostat kept in a glass dewar. For thermoelectric power measurements, a thick specimen (nearly 5 mm) was kept in good thermal contact between two copper blocks. The two copper blocks were again separated by two fused quartz rods and a low power heater was arranged around the upper copper block to maintain a temperature difference. Copper-constantan thermocouples were used to monitor  $\Delta T$  between the two ends of the specimen. The thermo-induced voltage,  $\Delta V$ , was read with a digital micro-voltmeter with a sensitivity of 0 to  $1 \mu V$ .

#### **3. Results**

Fig. 1 shows the X-ray diffractograms (XRD) of undoped and fluorine-doped GdBa<sub>2</sub>Cu<sub>3</sub>O<sub>7-x</sub> for (x = 0, 0.05 and 0.1) samples. The XRD of undoped material shows that it has a single orthorhombic phase. The



*Figure 1* X-ray diffraction patterns of (a)  $GdBa_2Cu_3O_7$ , (b)  $GdBa_2Cu_3O_{6.95}F_{0.05}$ , and (c)  $GdBa_2Cu_3O_{6.90}F_{0.10}$ .

positions and intensities of the peaks are in good agreement with the reported values. The fluorinedoped samples, the intensity and position of the main peaks are found to change. In an undoped sample, the intense peak is obtained at  $2\theta = 33^{\circ}$ , whereas its intensity decreased and shifted to 32.4° in  $x = 0.1$ fluorine-doped sample. Additional peaks were also obtained at  $2\theta$  equal to  $29.8^\circ$ ,  $30.2^\circ$ ,  $35.8^\circ$ ,  $45.2^\circ$  and 49°. These peaks which were only just visible in the  $x = 0.05$  fluorine-doped sample developed into welldefined peaks in the  $x = 0.1$  doped sample. The decrease in the intensity of the main peak and the presence of additional peaks in the fluorine-doped samples suggest that additional phases have formed due to fluorine doping. The XRD of  $GdF_3$  was also taken and a comparison showed that the additional peaks did not correspond to the  $GdF_3$  phase.

The resistance of the samples has been studied as a function of temperature for the three samples  $(x =$ 0.0, 0.05 and 0.1) and the results are shown in Fig. 2. The superconducting transition was observed in all three samples above LNT but  $T_c(0)$  decreased with

fluorine concentration. The observed  $T_c(ON)$  and  $T<sub>c</sub>(0)$  of undoped and doped samples are given in Table I. The thermoelectric power (TEP) of three samples ( $x = 0.0, 0.05$  and 0.1) as a function of temperature is shown in Figs 3a to c, respectively. The TEP drops to zero in all three samples. The transition temperatures obtained from these results are given in Table I. The graphs show that (i) the transition temperature decreases with fluorine doping, (ii) the value of thermopower at 220 K is almost the same  $(1.5 \,\mu\text{V})$  $K^{-1}$ ) in all three compounds, (iii) with decreasing temperature the thermopower decreases linearly until 120 K in undoped and 180K in fluorine-doped compounds, (iv) below  $120K$  in undoped and  $180K$  in doped samples the thermopower is enhanced and reaches a maximum value ( $S = 0.8 \mu$ V K<sup>-1</sup> for  $x = 0$ ,  $S = 3.25 \,\mu\text{V K}^{-1}$  for  $x = 0.05$  and  $S = 3.5 \,\mu\text{V K}^{-1}$  for  $x = 0.1$ ) just below the transition temperature.

#### **4. Discussion**

The resistivity and thermopower data presented in Figs 2 and 3 show that  $T_c$  decreases with increasing



*Figure 2* Resistance variation with temperature for (a) (o)  $GdBa_2Cu_3O_7,\hspace{1mm} \text{(b)} \hspace{1mm} \text{(c)} \hspace{1mm} GdBa_2Cu_3O_{6.95}F_{0.05}, \hspace{1mm} \text{and} \hspace{1mm} \text{(c)} \hspace{1mm} \text{(A)}$  $GdBa<sub>2</sub>Cu<sub>3</sub>O<sub>6.90</sub>F<sub>0.10</sub>.$ 



fluorine concentration. While preparing the fluorinedoped compound, the  $Gd_2O_3$  concentration was reduced and  $GdF_3$  was added in its place, in order to maintain the same gadolonium composition in all the compounds. Thus in the starting material itself, oxygen concentration is less. It has been established by several workers that  $T<sub>c</sub>$  is very sensitive to oxygen concentration and a decrease in oxygen concentration could lead to a decrease in  $T_c$ . Further XRD pattens showed the presence of additional phases in fluorine-doped samples. It has also been found that  $GdF_3$  phases are not present and these additional phases probably could be due to the formation of some oxy-fluorides. If oxy-fluorides are formed, the effective concentration of oxygen present in the Cu-O phases could decrease and this may lead to a decrease in  $T_c$ .

The thermoelectric power of  $YBa_2Cu_3O_7$  has been measured by a number of groups [7-10]. Similar to our results, they also observed an enhancement of the thermopower just above the superconducting transition temperature. Uher and Kaiser [8] suggested it to be a phonon drag truncated by the onset of superconductivity. Jha *et al.* [10] have also observed similar behaviour in  $YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7</sub>$  cooled at different rates. This behaviour is very similar to the phonon drag peak expected at a temperature about  $\theta_{\rm D}/5$  where  $\theta_{\rm D}$  is the Debye temperature. Taking a reasonable value of  $\theta_{\rm D}$ as  $400 \text{ K}$  [11] the observed peak at its  $1/T$ -like fall are consistent with the phonon drag mechanism. The phonon drag mechanism should, in general, be reduced by alloying, due to an increase in the phonon scattering. However, Srinivasan *et al.* [9], who studied the thermopower of an alloy in which 20% yttrium was replaced by erbium, found no such reduction. Ausloos *et al.* [12] suggested another explanation which depends on the granular nature of these materials. Assuming a fraction of the grains to be superconducting, they were able to achieve the desired form of the temperature dependence of thermopower, S. Jha *et al.* [10] have suggested an alternating mechanism based on superconducting fluctuations in the normal phase, rather than a phonon drag effect. It appears that no single mechanism described above is actually responsible for the enhancement in a strongly interacting superconducting system like  $YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7</sub>$ , and we feel that more detailed studies are required before an exact mechanism could be postulated.

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*Figure 3* Thermoelectric power variation with temperature for (a)  $GdBa_2Cu_3O_7$ , (b)  $GdBa_2Cu_3O_{6.95}F_{0.05}$ , and (c)  $GdBa_2Cu_3O_{6.90}F_{0.10}$ 

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